

## Inhibitory and enhancing effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity: Dependence on the concentrations of NO and H<sub>2</sub>O<sub>2</sub>

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Accepted by Professor M. Davies

(Received 9 July 2006; in revised form 5 October 2006)

### Abstract

Nitric oxide (NO) has been shown to both enhance hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) toxicity and protect cells against H<sub>2</sub>O<sub>2</sub> toxicity. In order to resolve this apparent contradiction, we here studied the effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity in cultured liver endothelial cells over a wide range of NO and H<sub>2</sub>O<sub>2</sub> concentrations. NO was generated by spermine NONOate (SpNO, 0.001–1 mM), H<sub>2</sub>O<sub>2</sub> was generated continuously by glucose/glucose oxidase (GOD, 20–300 U/l), or added as a bolus (200 μM). SpNO concentrations between 0.01 and 0.1 mM provided protection against H<sub>2</sub>O<sub>2</sub>-induced cell death. SpNO concentrations > 0.1 mM were injurious with low H<sub>2</sub>O<sub>2</sub> concentrations, but protective at high H<sub>2</sub>O<sub>2</sub> concentrations. Protection appeared to be mainly due to inhibition of lipid peroxidation, for which SpNO concentrations as low as 0.01 mM were sufficient. SpNO in high concentration (1 mM) consistently raised H<sub>2</sub>O<sub>2</sub> steady-state levels in line with inhibition of H<sub>2</sub>O<sub>2</sub> degradation. Thus, the overall effect of NO on H<sub>2</sub>O<sub>2</sub> toxicity can be switched within the same cellular model, with protection being predominant at low NO and high H<sub>2</sub>O<sub>2</sub> levels and enhancement being predominant with high NO and low H<sub>2</sub>O<sub>2</sub> levels.

**Keywords:** Nitrogen monoxide, nitric oxide, hydrogen peroxide, catalase, glutathione peroxidase, lipid peroxidation

**Abbreviations:** 8-Br-cGMP, 8-bromoguanosine-3',5'-cGMP; cGMP, cyclic guanosine monophosphate; GOD, glucose oxidase; H<sub>2</sub>O<sub>2</sub>, hydrogen peroxide; KH buffer, Krebs-Henseleit buffer; LDH, lactate dehydrogenase; NO, nitric oxide; NO<sub>x</sub>, reactive nitrogen oxide species; PKG, cGMP-dependent protein kinase; SpNO, spermine NONOate; TBARS, thiobarbituric acid - reactive substances

### Introduction

It is well known that nitric oxide (NO) can elicit cell and tissue injury [1–5]. Beside NO itself, oxidation products of NO such as NO<sub>2</sub> or N<sub>2</sub>O<sub>3</sub> and/or an interplay with reactive oxygen species (e.g. O<sub>2</sub><sup>•-</sup>, H<sub>2</sub>O<sub>2</sub>) contribute to injury [3,6,7]. As one of the interplays, co-operation of NO and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) in their cytotoxicity against different cell types including rat liver endothelial cells, HepG2 cells, murine lymphoma cells, rabbit gastric mucosal cells, embryonic chick cardiomyocytes and a human epithelial ovarian cancer cell line has been described

by several groups [7–13]. In recent experiments with isolated rat liver endothelial cells and hepatocytes, we have shown that NO inhibits H<sub>2</sub>O<sub>2</sub> degradation by action on both H<sub>2</sub>O<sub>2</sub>-degrading pathways, catalase and the glutathione peroxidase pathway, and that this inhibition of H<sub>2</sub>O<sub>2</sub> degradation can account for the co-operative action of NO and H<sub>2</sub>O<sub>2</sub> (manuscript submitted for publication). Catalase was inhibited by NO itself, glutathione peroxidase by reactive nitrogen oxide species (NO<sub>x</sub>), and the relative contribution of both pathways depended on both, the cell type and the H<sub>2</sub>O<sub>2</sub> levels applied.

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Besides its injurious actions, NO has also been described to be cytoprotective in numerous *in vivo* and *in vitro* studies [2,3,14–19]. Along this line, protective effects of NO against H<sub>2</sub>O<sub>2</sub> toxicity have also been reported in endothelial cells [20–23]. Although the differences in the effects observed—i.e. co-operative cytotoxic effects vs. cytoprotection—might be dependent on the cell type employed, they might also be dependent on experimental conditions such as concentrations of the reactive species applied etc. As comparative studies have not yet been done, we here set out to assess whether the occurrence of either enhancing or inhibitory effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity is dependent on NO and/or H<sub>2</sub>O<sub>2</sub> concentrations.

## Materials and methods

### Materials

Horseradish peroxidase (grade I) and glucose oxidase (GOD, from *Aspergillus niger*) were purchased from Roche Molecular Biochemicals (Mannheim, Germany). Scopoletin (7-hydroxy-6-methoxy-2H-1-benzopyran-2-one), 8-bromoguanosine-3',5'-cGMP (8-Br-cGMP), KT5823,  $\alpha$ -tocopherol and hydrogen peroxide solution were obtained from Sigma-Aldrich (Taufkirchen, Germany). Spermine NONOate (SpNO) was from Situs (Düsseldorf, Germany), RPMI 1640 medium from Gibco (Karlsruhe, Germany) and C11-BODIPY<sup>®581/591</sup> (4,4-difluoro-5-(4-phenyl-1,3-butadienyl)-4-bora-3a,4a-diaza-s-indacene-3-undecanoic acid) from Molecular Probes (Leiden, The Netherlands). All other chemicals used were purchased from Merck (Darmstadt, Germany).

### Cell isolation and culture

A rat liver endothelial cell line, derived from the liver of a male Wistar rat, was used for the experiments. The cells had been isolated and characterized as described previously [24]. Cells were cultured in RPMI 1640 medium supplemented with fetal calf serum (20%), L-glutamine (2 mM), penicillin/streptomycin (50 U/ml and 50  $\mu$ g/ml, respectively) and dexamethasone (1  $\mu$ M). Subcultures were obtained by trypsinization (split ratio 1:3), and cells were used for experiments on day 6 or 7 after subcultivation.

### Incubation procedure

At the beginning of the experiments, cells in 12.5 cm<sup>2</sup> culture flasks (BD Biosciences, Heidelberg, Germany) were washed three times with Hanks' balanced salt solution and then covered with 2.5 ml of modified Krebs–Henseleit buffer (KH: NaCl 115 mmol/l; NaHCO<sub>3</sub> 25 mmol/l; KCl 5.9 mmol/l; MgCl<sub>2</sub> 1.2 mmol/l; NaH<sub>2</sub>PO<sub>4</sub> 1.2 mmol/l; Na<sub>2</sub>SO<sub>4</sub> 1.2 mmol/l;

CaCl<sub>2</sub> 2.5 mmol/l, Hepes 20 mmol/l; pH 7.4, supplemented with 10 mM glucose). All experiments were performed at 37°C. The NO donor SpNO was added to the incubation solution 15 min prior to addition of H<sub>2</sub>O<sub>2</sub> or of GOD. Alternatively, cells were preincubated for 30 min with 8-Br-cGMP or KT5823; solvent controls were included. In other experiments, cells were preincubated with  $\alpha$ -tocopherol as described by Martin et al. [25]. Briefly, cells were incubated for 20 h in cell culture medium containing  $\alpha$ -tocopherol (added from a stock solution, 10 mg/ml in ethanol) in a final concentration of 60  $\mu$ M;  $\alpha$ -tocopherol was not added to the solutions during the experiments.

### Assessment of cell injury

Lactate dehydrogenase (LDH) release was used to indicate occurrence of lethal cell injury [26].

### Determination of H<sub>2</sub>O<sub>2</sub> steady-state levels

H<sub>2</sub>O<sub>2</sub> concentrations in the incubation medium were determined fluorimetrically by monitoring the horseradish peroxidase-dependent oxidation of scopoletin, using the assay conditions described previously (Ref. [27] for low H<sub>2</sub>O<sub>2</sub> concentrations, Ref. [28] for higher H<sub>2</sub>O<sub>2</sub> concentrations).

### Determination of nitric oxide steady-state levels

NO concentrations in cell culture wells were determined with an electrochemical probe (ISO-NO electrochemical probe from World Precision Instruments, Berlin, Germany) as described in Ref. [29].

### Determination of lipid peroxidation

Lipid peroxidation was assessed by the determination of thiobarbituric acid-reactive substances (TBARS) and by following the oxidation of the fluorescent reporter molecule C11-BODIPY<sup>®581/591</sup>. TBARS were determined in the supernatant incubation solution after various incubation times as described previously [30]. In a second series, endothelial cells cultured on 12-well plates were loaded with the fluorescent, oxidation-sensitive fatty acid analogue C11-BODIPY<sup>®581/591</sup> (10  $\mu$ M in KH buffer, 60 min loading at 37°C) [31,32], washed and then incubated at 37°C in dye-free, bicarbonate-free, Hepes-buffered modified KH buffer (supplemented with 10 mM D-glucose) in a fluorescence microplate reader (FLUOstar OPTIMA, BMG Labtech, Offenburg, Germany). To part of the incubations, SpNO was added in a final concentration of 0.1 mM or 10  $\mu$ M. Fluorescence was recorded at  $\lambda_{\text{exc}} = 485$  nm,  $\lambda_{\text{em}} = 520$  nm and  $\lambda_{\text{exc}} = 560$  nm,  $\lambda_{\text{em}} = 590$  nm in 60 s intervals using the top reading mode. After 15 min, GOD was added

in a final activity of 300 U/l, and fluorescence recordings were continued for 3 h. Values are given as a ratio of the intensity of the green fluorescence of the oxidized indicator ( $\lambda_{exc} = 485 \text{ nm}$ ,  $\lambda_{em} = 520 \text{ nm}$ ) to the intensity of the red fluorescence of the non-oxidized indicator ( $\lambda_{exc} = 560 \text{ nm}$ ,  $\lambda_{em} = 590 \text{ nm}$ ).

### Statistics

All experiments were performed in duplicate and repeated 3–11 times. Data are expressed as means  $\pm$  SD. Data obtained from two groups were compared by means of Student's *t*-test and comparisons among multiple groups were performed using an analysis of variance with Student-Newman-Keuls, Dunnett or Bonferroni post hoc comparisons, as appropriate. A *p* value of  $<0.05$  was considered significant.

## Results

### Cytotoxicity of hydrogen peroxide

When confluent cultures of liver endothelial cells were incubated in KH buffer with the hydrogen peroxide-generating system glucose (10 mM)/GOD, a GOD activity of 20 U/l did not cause any injury (Figure 1A), 50 U/l caused slight injury (Figure 1B), and higher GOD activities, i.e. 150 U/l (LDH release  $43 \pm 38\%$  after 3 h) or 300 U/l (Figure 1C), led to a more pronounced and more rapid loss of viability.

When  $\text{H}_2\text{O}_2$  was not generated continuously by the enzymatic system but authentic  $\text{H}_2\text{O}_2$  was added as a bolus (200  $\mu\text{M}$ ), rapid and marked endothelial cell injury occurred (Figure 2).

### Cytotoxicity of NO

Addition of 1 mM of the NO donor SpNO did not cause any cell injury over 3 h of incubation in the glucose-containing KH buffer ( $3 \pm 1\%$  LDH release after 3 h), but caused moderate injury thereafter ( $27 \pm 21\%$  after 5 h and  $67 \pm 19\%$  after 7 h). SpNO at a concentration of 0.5 mM showed lower toxicity and SpNO at a concentration of 0.1 mM no toxicity at all ( $4 \pm 1\%$  after 7 h).

### Enhancing and protective effects of NO on the toxicity of hydrogen peroxide

Although 20 U/l GOD were not toxic on their own, the additional administration of SpNO enhanced  $\text{H}_2\text{O}_2$  toxicity (Figure 1A), in line with the co-operative effect of NO and  $\text{H}_2\text{O}_2$  we described previously [7,8]. This enhancement of  $\text{H}_2\text{O}_2$  toxicity by NO was dependent on the concentration of the NO donor: 0.1 mM SpNO only showed a tendency towards enhancement of  $\text{H}_2\text{O}_2$  toxicity, 0.5 mM SpNO markedly increased  $\text{H}_2\text{O}_2$  toxicity and 1 mM SpNO had an

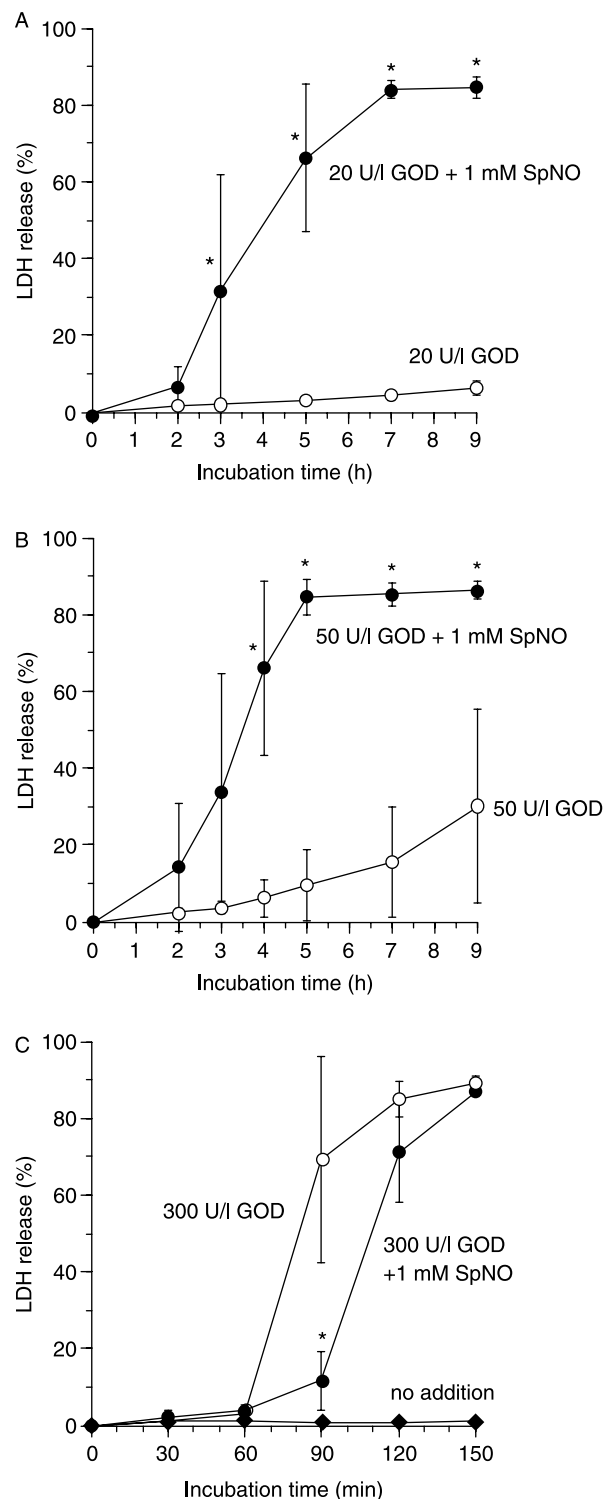


Figure 1. Enhancing and inhibitory effects of the NO donor SpNO on hydrogen peroxide toxicity to liver endothelial cells exposed to a hydrogen peroxide-generating system. Cultured rat liver endothelial cells were incubated with the hydrogen peroxide-generating system glucose/GOD and/or the NO donor SpNO in KH buffer. To this end, cells were preincubated or not with 1 mM SpNO for 15 min (in KH buffer containing 10 mM D-glucose), then (time "0"), GOD was added to this medium in a final activity of 20 U/l (A), 50 U/l (B) or 300 U/l (C). Cell injury was assessed by the release of LDH. Data are means  $\pm$  SD of 3–5 experiments. \*Significantly different to the respective incubation without SpNO,  $p < 0.05$ .

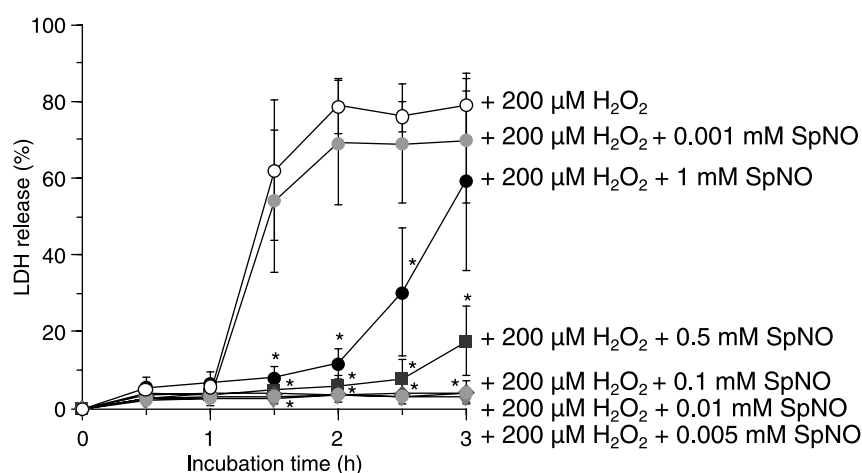


Figure 2. Effects of different SpNO concentrations on the endothelial cell injury elicited by a bolus of hydrogen peroxide. Cultured liver endothelial cells were preincubated or not with different concentrations of the NO donor SpNO in KH buffer containing 10 mM D-glucose for 15 min. Then (time “0”), a bolus of authentic H<sub>2</sub>O<sub>2</sub> (final concentration 200 μM) was added. Cell injury was assessed by the release of LDH. Data are means ± S.D. of 4 experiments. \* Significantly different to the respective incubation without SpNO,  $p < 0.05$ .

even more pronounced effect (Figure 3A). Similar enhancement of H<sub>2</sub>O<sub>2</sub> toxicity by 1 mM SpNO was observed when 50 U/l GOD was used (Figure 1B). Again, the enhancement was less pronounced with 0.5 mM SpNO and only marginal with 0.1 mM SpNO (data not shown).

At an intermediate GOD activity, i.e. 150 U/l GOD, however, 1 mM SpNO only increased H<sub>2</sub>O<sub>2</sub> toxicity in 5 out of 11 experiments, and was protective in six experiments. SpNO in a concentration of 0.1 mM, however, consistently provided protection against the injury induced by 150 U/l GOD, even in those experiments where 1 mM SpNO enhanced cell injury (data not shown). When cells were exposed to 300 U/l GOD, 1 mM SpNO did not cause any aggravation of injury but consistently provided protection (Figure 1C). Similarly, 0.1 mM SpNO or even 10 μM SpNO provided protection (Figure 3B).

As the time course and extent of endothelial cell injury induced by the “intermediate” GOD activity of 150 U/l showed a relative high variability, we grouped these experiments according to the time course/extent of the injury (LDH release after 3 h < 40% or > 40%), and surprisingly, in all experiments with a slow time course/smaller extent of the injury ( $n = 5$ ) 1 mM SpNO strongly enhanced H<sub>2</sub>O<sub>2</sub>-induced cell injury, while in all experiments with a more rapid H<sub>2</sub>O<sub>2</sub>-induced cell injury 1 mM SpNO was protective ( $n = 6$ ).

When H<sub>2</sub>O<sub>2</sub> was applied as a bolus, 1 mM SpNO was also consistently protective, although the protection was only partial (Figure 2). The protection by SpNO was enhanced at lower SpNO concentrations, i.e. 0.5 mM SpNO provided almost complete and 0.1 mM SpNO provided complete protection. Even concentrations as low as 5 and 10 μM SpNO were

fully protective and only at 1 μM SpNO, the protective effect was lost.

Thus, it proved to be dependent on both, SpNO concentrations and intensity of H<sub>2</sub>O<sub>2</sub>-induced cell injury whether SpNO aggravated H<sub>2</sub>O<sub>2</sub>-induced cell injury or protected against H<sub>2</sub>O<sub>2</sub>-induced cell injury, with aggravation being predominant at high SpNO concentrations (> 0.1 mM) and low/lower H<sub>2</sub>O<sub>2</sub>-induced cell injury (20–50 U/l GOD, less sensitive cell passages with 150 U/l GOD) and protection being predominant at low SpNO concentrations (< 0.1 mM) and high/rapid H<sub>2</sub>O<sub>2</sub>-induced cell injury (150 U/l GOD in the more sensitive cultures, 300 U/l GOD, H<sub>2</sub>O<sub>2</sub> bolus).

#### Steady-state levels of NO

For the experimental setting used in the current study, the steady-state levels of NO (in the incubation medium) are shown in Figure 4 for the borderline SpNO concentration of 0.1 mM, i.e. the SpNO concentration on the verge between protection and enhancement of H<sub>2</sub>O<sub>2</sub> toxicity. 0.1 mM SpNO gave peak NO concentrations of 1–2 μM NO.

#### Hydrogen peroxide steady-state levels

In liver endothelial cell cultures incubated in KH buffer (supplemented with 10 mM glucose) in the absence of GOD, H<sub>2</sub>O<sub>2</sub> steady-state levels were below 0.3 μM. The addition of 20 U/l GOD led to an increase in H<sub>2</sub>O<sub>2</sub> steady-state levels that reached a maximum at 1–2 h of incubation (Figure 5A), levels at 1 h being  $3.2 \pm 1.5 \mu\text{M}$ . When 50 U/l GOD were applied, H<sub>2</sub>O<sub>2</sub> steady-state levels rose higher, reaching  $13.9 \pm 6.2 \mu\text{M}$  after 1 h of incubation, and remained at this level for at

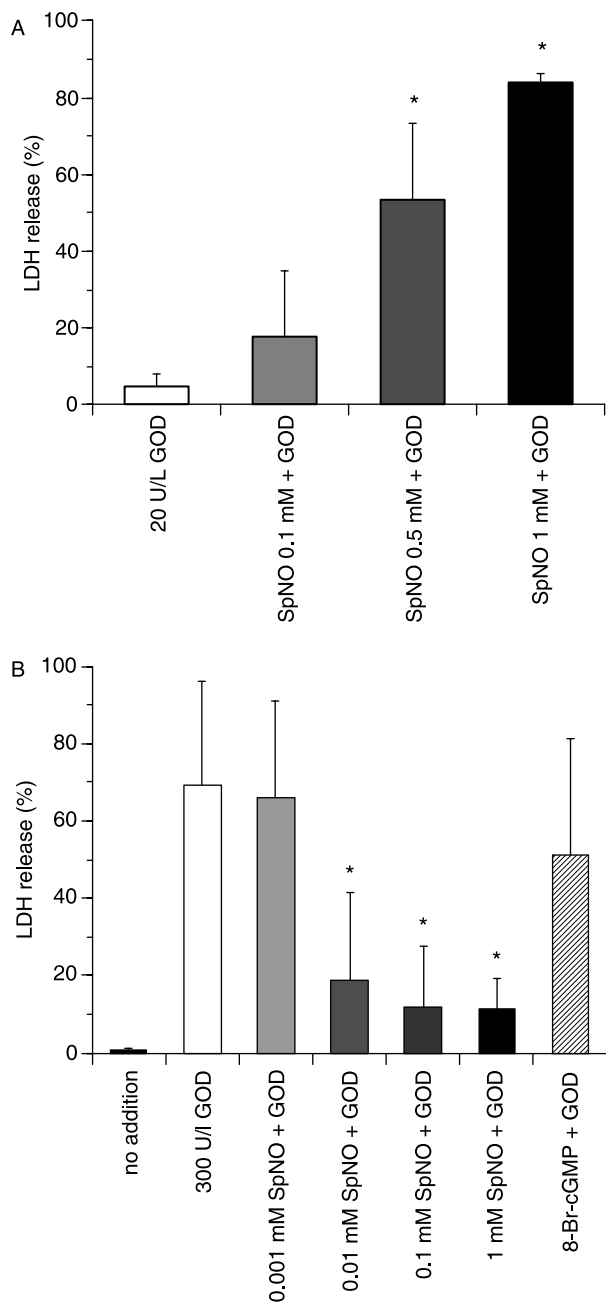


Figure 3. Effects of different SpNO concentrations on hydrogen peroxide toxicity to liver endothelial cells exposed to a hydrogen peroxide-generating system. Cultured liver endothelial cells were preincubated or not with different concentrations of the NO donor SpNO in KH buffer containing 10 mM D-glucose for 15 min, then (time “0”), GOD was added to this medium in a final activity of 20 U/l (A) or 300 U/l (B). In order to assess whether guanylate cyclase might be involved in cytoprotection by NO, additional cultures were preincubated with 8-Br-cGMP (50  $\mu$ M, 30 min preincubation) instead of the NO donor (B). Cell injury was assessed after 7 h (A) or after 90 min (B) by the release of LDH. Data are means  $\pm$  S.D. of 3 experiments. \* Significantly different to the respective incubation with GOD alone (without SpNO/8-Br-cGMP),  $p < 0.05$ .

least 4 h (values after 5 h of incubation:  $12.4 \pm 9.6 \mu$ M). In the presence of 150 U/l GOD,  $H_2O_2$  steady-state levels rose to  $40.3 \pm 9.8 \mu$ M after 2 h of incubation, and with 300 U/l GOD,  $H_2O_2$  steady-state levels after 2 h

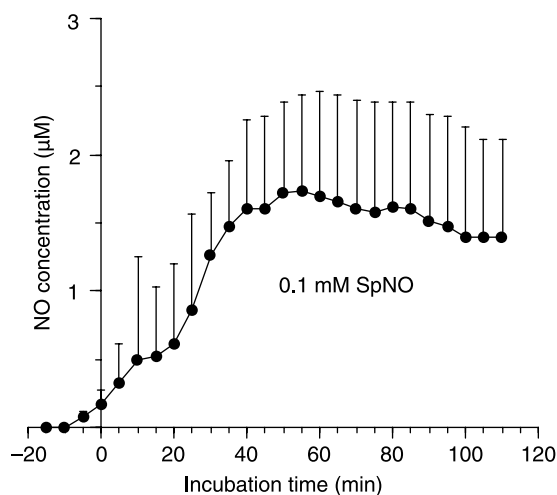


Figure 4. NO steady-state levels in liver endothelial cell cultures exposed to 0.1 mM of the NO donor SpNO. NO steady-state levels in liver endothelial cell cultures incubated with 0.1 mM SpNO in KH buffer (supplemented with 10 mM D-glucose) at 21%  $O_2$  (as in the experiments shown in Figures 1–3, 5–9) were monitored continuously with an NO-sensitive electrode. To facilitate comparison with the other figures, the usual preincubation period is shown in the negative range of the x-axis (i.e. time “0” equals the time point when GOD or  $H_2O_2$  was added in the experiments shown in Figures 1–3 and 5). Data are means  $\pm$  S.D. of 4 experiments.

were  $89.4 \pm 2.9 \mu$ M. A bolus of 200  $\mu$ M  $H_2O_2$  has previously been shown to be degraded by the same cell type in the same experimental system to concentrations below 100  $\mu$ M within 10 min and below 20  $\mu$ M after about 36 min (manuscript submitted for publication).

#### Effects of NO/ $NO_x$ on hydrogen peroxide steady-state levels

As described previously, SpNO (1 mM) added in addition to 20 U/l GOD did not affect  $H_2O_2$  levels up to 1 h, but increased  $H_2O_2$  levels thereafter (Figure 5A). In contrast to 1 mM SpNO—and in line with its only marginal effect on viability (Figure 3A)—0.1 mM SpNO did not alter  $H_2O_2$  steady-state levels significantly (Figure 5A).

Also at higher GOD activities, i.e. in the range where protective effects by SpNO were observed, the addition of 1 mM SpNO strongly raised  $H_2O_2$  steady-state levels (Figure 5B,C). That SpNO (1 mM) also inhibits the degradation of a 200  $\mu$ M  $H_2O_2$  bolus by liver endothelial cells has already been shown in the same system (manuscript submitted for publication).

#### No evidence for an involvement of a cGMP-dependent pathway in the protection by NO against $H_2O_2$

A major action of NO is the activation of guanylate cyclase with subsequent formation of cyclic guanosine monophosphate (cGMP). Protective effects of cGMP have been reported in several models of cell injury

[14,18,19], and have, among others, been attributed to activation of cGMP-dependent protein kinase (PKG) [14,19]. Therefore, we tested whether the protective effect of SpNO can be mimicked by 8-Br-cGMP, a soluble and membrane-permeable cGMP analogue. Preincubating the cells with 8-Br-cGMP (50 μM) for 30 min prior to the addition of GOD or of

a bolus of H<sub>2</sub>O<sub>2</sub>, a protocol that provided protection in other models of cell injury [19], did neither protect liver endothelial cells against the injury elicited by glucose/GOD (300 U/l; Figure 3B) nor against the injury elicited by a bolus of authentic H<sub>2</sub>O<sub>2</sub> (200 μM; Figure 6). Along the same line, an inhibitor of PKG, KT5823 (5 μM), did not prevent NO-mediated protection of endothelial cells against H<sub>2</sub>O<sub>2</sub>-induced injury (Figure 6). Thus, protection by NO against H<sub>2</sub>O<sub>2</sub>-induced injury appears to be independent of cGMP and PKG.

*Effect of NO on lipid peroxidation*

NO has also been described to inhibit lipid peroxidation [15,16]; therefore, we tested this possibility next. H<sub>2</sub>O<sub>2</sub> application, whether by the continuously H<sub>2</sub>O<sub>2</sub>-generating system glucose/GOD (data not shown) or by bolus addition (Figure 7), initiated lipid peroxidation as detected by the formation of TBARS.

The H<sub>2</sub>O<sub>2</sub>-induced formation of TBARS was inhibited by SpNO—strongly with SpNO concentrations of 1 mM but already (and even slightly better) with SpNO concentrations of 0.1 mM or 10 μM. This was confirmed using a fluorescent reporter molecule for lipid peroxidation, the fluorescent fatty acid analogue C11-BODIPY<sup>®581/591</sup>, the fluorescence of which shifts from red to green upon oxidation of its conjugated double bond [31,32]. After addition of 300 U/l GOD, the green/red fluorescence ratio of C11-BODIPY<sup>®581/591</sup> increased, indicating oxidation of this indicator (Figure 8), and 0.1 mM SpNO (Figure 8) or 10 μM SpNO (data not shown) completely inhibited this oxidation (in the presence of the NO donor, the fluorescence ratio remained even below the values of control cells not exposed to exogenous H<sub>2</sub>O<sub>2</sub>).

This inhibitory effect on lipid peroxidation was thus not only present, but also occurred at similar concentrations as the protective effect of NO—and thus appears to be a likely explanation for the NO-induced protection observed. This was reinforced by the findings that preincubating the cells with the

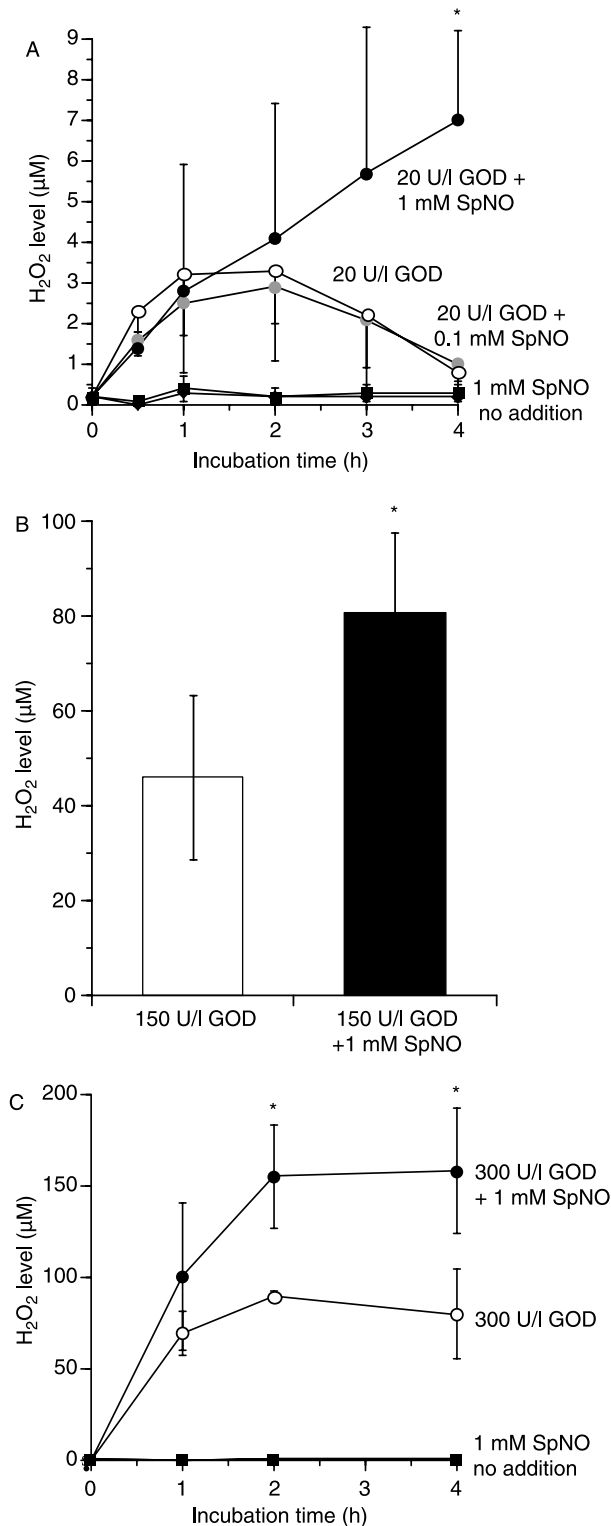


Figure 5. Increase in hydrogen peroxide steady-state levels in endothelial cell cultures in the presence of higher concentrations of the NO donor SpNO. Cultured liver endothelial cells were preincubated or not with different concentrations of the NO donor SpNO in KH buffer containing 10 mM D-glucose for 15 min. Then (time “0”), GOD was added to this medium in a final activity of 20 U/l (A), 150 U/l (B) or 300 U/l (C). Hydrogen peroxide steady-state levels were assessed in samples taken at different time points (B: 150 min) using the horseradish peroxidase-catalyzed oxidation of scopoletin. Calibration was performed using authentic hydrogen peroxide. Data are means ± S.D. of 4–5 experiments. \* Significantly different to the respective incubation without SpNO, *p* < 0.05.

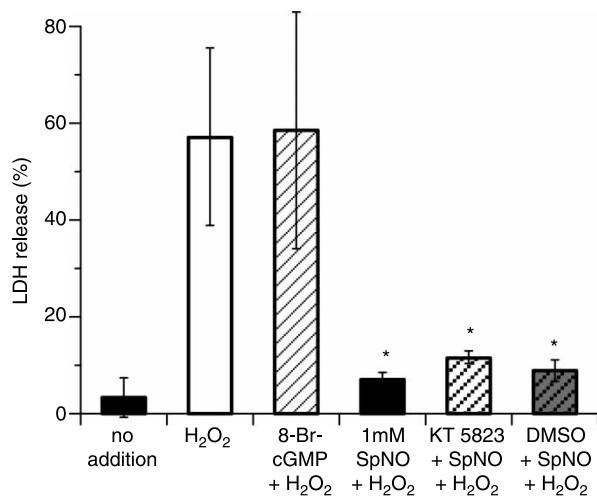


Figure 6. No evidence for an involvement of a cGMP-dependent pathway in the protective effect of NO against hydrogen peroxide toxicity. Cultured liver endothelial cells were preincubated with the NO donor SpNO (1mM), the cGMP analogue 8-Br-cGMP (50  $\mu$ M) and/or the PKG inhibitor KT5823 (5  $\mu$ M), or the solvent of the latter (DMSO), in KH buffer (supplemented by 10 mM D-glucose) for 15 min (SpNO) or 30 min (8-Br-cGMP, KT5823, solvent). Then, a bolus of authentic H<sub>2</sub>O<sub>2</sub> (final concentration 200  $\mu$ M) was added, and cells were incubated for 90 min. Cell injury was assessed by the release of LDH. Data are means  $\pm$  S.D. of 3 experiments. \* Significantly different to the respective incubation with H<sub>2</sub>O<sub>2</sub> alone,  $p < 0.05$ .

potent inhibitor of lipid peroxidation,  $\alpha$ -tocopherol, mimicked the protective effect of SpNO (Figure 9), while in  $\alpha$ -tocopherol-pretreated cells SpNO did not provide any further protection.

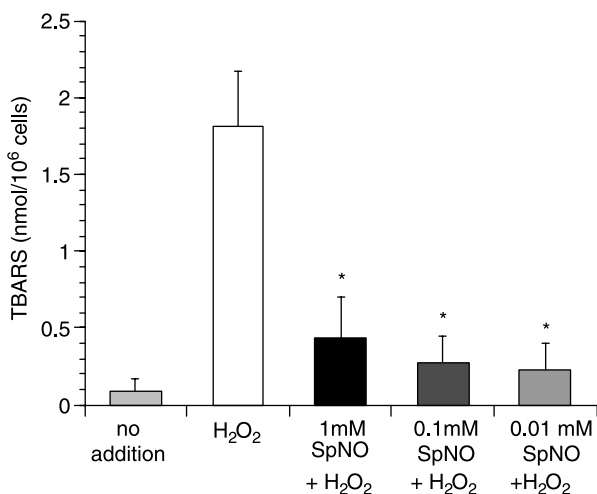


Figure 7. Inhibition of the formation of TBARS by the NO donor SpNO and its dependence on the SpNO concentration. Cultured liver endothelial cells were preincubated or not with different concentrations of the NO donor SpNO in KH buffer containing 10 mM D-glucose for 15 min. Then, a bolus of authentic H<sub>2</sub>O<sub>2</sub> (final concentration 200  $\mu$ M) was added. As markers of lipid peroxidation, TBARS were determined after 3 h of incubation. Data are means  $\pm$  S.D. of 4 experiments. \* Significantly different to the respective incubation without SpNO,  $p < 0.05$ .

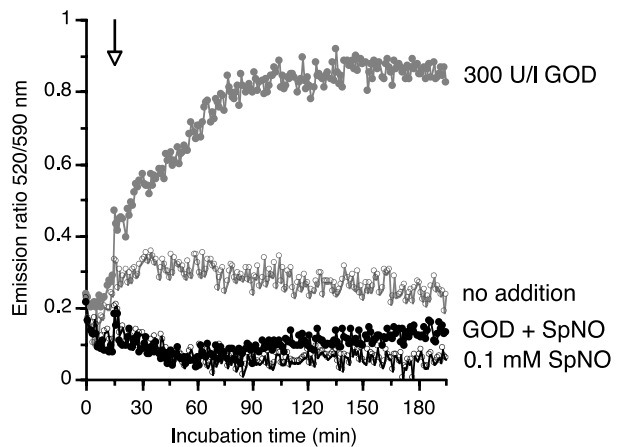


Figure 8. Inhibition of lipid peroxidation by the NO donor SpNO. Liver endothelial cells cultured on 12-well plates were loaded with the fluorescent, oxidation-sensitive fatty acid analogue C11-BODIPY<sup>581/591</sup> (10  $\mu$ M, 60 min), and then incubated at 37°C in modified KH buffer (supplemented with 10 mM D-glucose) in a fluorescence microplate reader. To part of the incubations, SpNO was added in a final concentration of 0.1 mM at time zero. Fluorescence was recorded at  $\lambda_{exc} = 485$  nm,  $\lambda_{em} = 520$  nm and  $\lambda_{exc} = 560$  nm,  $\lambda_{em} = 590$  nm. After 15 min, GOD was added to part of the wells in a final activity of 300 U/l (arrow), and fluorescence recordings were continued for 3 h. Values are given as a ratio of the intensity of the green fluorescence of the oxidized indicator ( $\lambda_{exc} = 485$  nm,  $\lambda_{em} = 520$  nm) to the intensity of the red fluorescence of the non-oxidized indicator ( $\lambda_{exc} = 560$  nm,  $\lambda_{em} = 590$  nm). Data shown represent a representative recording from one of five experiments performed in duplicates.

## Discussion

We have previously shown that NO and H<sub>2</sub>O<sub>2</sub> act cooperatively in their cytotoxic action on liver endothelial cells, and that this co-operativity is due to inhibition of H<sub>2</sub>O<sub>2</sub> degradation by NO (manuscript submitted for publication). However, protective effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity have also been described, and here we could show a switch between the enhancing and a protective effect of NO on H<sub>2</sub>O<sub>2</sub> toxicity in the same cellular system—solely dependent on NO and H<sub>2</sub>O<sub>2</sub> levels (Figure 10). In general, the enhancing effect of SpNO occurred when toxicity by H<sub>2</sub>O<sub>2</sub> alone was relatively low and NO concentration high (NO concentration  $> 1-2$   $\mu$ M), and the protective effect predominated when toxicity by H<sub>2</sub>O<sub>2</sub> alone was high and/or the concentration of NO was low (i.e. below  $1-2$   $\mu$ M).

The enhancement of H<sub>2</sub>O<sub>2</sub> toxicity with high NO concentrations (the SpNO concentration of 0.5 mM has previously given an NO steady-state concentration around 3  $\mu$ M in the same cell culture model) [29] is in line with previous observations showing that inhibition of glutathione peroxidase required SpNO concentrations in excess of 0.1 mM, with marked inhibition being observed at 1 mM SpNO (around 10  $\mu$ M NO; manuscript submitted for publication). Inhibition of glutathione peroxidase appeared to be caused by

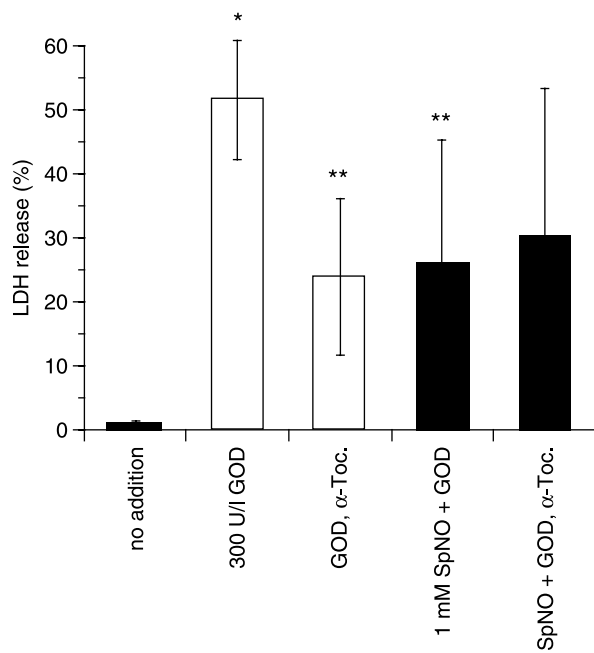


Figure 9. α-Tocopherol preincubation mimics the protective effect of SpNO on hydrogen peroxide toxicity to liver endothelial cells. Cultured liver endothelial cells were preincubated or not with α-tocopherol (α-Toc.; 60 μM in cell culture medium, 20 h) and then treated or not with the NO donor SpNO (1 mM) in KH buffer containing 10 mM D-glucose for 15 min. Thereafter (time “0”), GOD was added in a final activity of 300 U/l. Cell injury was assessed after 120 min by the release of LDH. Data are means ± SD of four experiments. \*Significantly different to the respective incubation without GOD, *p* < 0.05. \*\*Significantly different to the respective incubation with GOD alone (without SpNO/α-tocopherol), *p* < 0.05.

oxidation products of NO, reactive nitrogen oxide species (NO<sub>x</sub>), not by NO itself, and was therefore a delayed effect occurring after about 2 h of NO exposure (Figure 10). Inhibition of catalase also occurs in the presence of NO. In contrast to inhibition of glutathione peroxidase, inhibition of catalase is caused by NO itself [10,33], and is therefore an immediate—and reversible—effect. It required similar NO concentrations as inhibition of glutathione peroxidase, however, it appeared to contribute to the decreased H<sub>2</sub>O<sub>2</sub> degradation of liver endothelial cells confronted with NO only at high H<sub>2</sub>O<sub>2</sub> levels (manuscript submitted for publication), a finding that is in line with the predominant role of glutathione peroxidase in H<sub>2</sub>O<sub>2</sub> degradation of endothelial cells in the lower H<sub>2</sub>O<sub>2</sub> concentration range [34–36]. In the present study, these inhibitory effects of high concentrations of NO on H<sub>2</sub>O<sub>2</sub> degradation—whether by inhibition of glutathione peroxidase or of catalase—appeared to occur throughout, irrespective of the concentration of H<sub>2</sub>O<sub>2</sub> applied, even under conditions where high levels of NO proved to be protective (Figure 5A–C).

The most likely mechanism for the protective effect of NO is an inhibition of lipid peroxidation by NO, as has been described previously [13,15,16]. H<sub>2</sub>O<sub>2</sub>-induced cell injury is mediated, besides by a calcium-dependent pathway, by an iron-dependent pathway that results in hydroxyl radical formation and subsequent lipid peroxidation (Figure 10) [28,37,38]. In the presence of glucose [37], and, in fact, in liver endothelial cells under most conditions (unpublished results), the iron-dependent pathway is the predominant one, and lipid peroxidation occurs

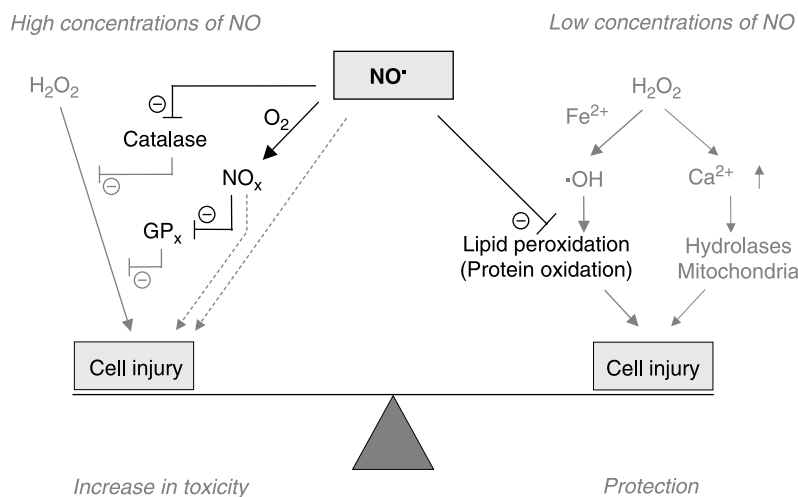


Figure 10. Enhancing and protective effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity. NO inhibits the H<sub>2</sub>O<sub>2</sub>-degrading enzyme catalase, and oxidation products of NO, NO<sub>x</sub>, inhibit the H<sub>2</sub>O<sub>2</sub>-degrading enzyme glutathione peroxidase (GP<sub>x</sub>). By these ways, H<sub>2</sub>O<sub>2</sub> steady-state levels are increased, leading to enhancement of cell injury. On the other hand, NO inhibits lipid peroxidation and thus exerts protective effects. While for the latter effects low (submicromolar) NO concentrations are sufficient, the inhibition of H<sub>2</sub>O<sub>2</sub> degradation requires higher NO levels (> 1–2 μM). Thus, high NO concentrations cause enhancement of toxicity (unless H<sub>2</sub>O<sub>2</sub>-induced injury is severe and occurs rapidly, when protection by inhibition of lipid peroxidation can prevail), while low NO concentrations exert protection (mainly via inhibition of lipid peroxidation). For further details, refer to text.



(Figures 7 and 8). NO, itself a radical and being present in the lipid phase of the cellular membranes in far higher concentrations than in the aqueous phase of the cell culture system, is likely to combine with radical intermediates of lipid peroxidation, e.g. peroxy radicals, and thus to interrupt this chain reaction [15,16,39,40]. This effect apparently only requires fairly low NO concentrations (in the aqueous phase); the 10  $\mu$ M SpNO that were already fully protective in the current study yielded maximal NO steady-state levels of about 300 nM in the same cell culture model as used here [29]. This fits well with a recent report by Hummel et al. [40], showing that ferrous iron + dioxygen-induced lipid peroxidation in HL-60 and U937 cells was inhibited by NO steady-state levels >75 nM. These NO concentrations are far lower than those that are required for the inhibition of H<sub>2</sub>O<sub>2</sub> degradation. Thus, it is explainable that at low NO concentrations protection by NO was predominant (compare results with 150 U/l GOD). At high NO concentrations, both, inhibition of H<sub>2</sub>O<sub>2</sub> degradation (Figure 5) and inhibition of lipid peroxidation (Figure 7) occurred. In general, with high NO concentrations, enhancement of H<sub>2</sub>O<sub>2</sub>-induced cell injury was observed with lower H<sub>2</sub>O<sub>2</sub> levels and/or lower cell injury induced by H<sub>2</sub>O<sub>2</sub> alone. However, protection prevailed when high H<sub>2</sub>O<sub>2</sub> amounts were applied and H<sub>2</sub>O<sub>2</sub>-induced cell injury was rapid (and lipid peroxidation prominent). With "intermediate" H<sub>2</sub>O<sub>2</sub> levels as demonstrated by the results with 150 U/l GOD, protective and enhancing effects appeared to be in a delicate balance and the resulting effect was thus sometimes switched by minor alterations such as cell passage. In line with the finding that inhibition of lipid peroxidation by NO might account for a major part of the protective effect of NO (although we cannot rule out additional inhibitory effects of NO on protein oxidation), preincubation of the cells with  $\alpha$ -tocopherol yielded similar protection as NO, and in  $\alpha$ -tocopherol-pretreated cells NO did no longer exert (additional) cytoprotection (Figure 9).

In the model used here, i.e. acute exposure of endothelial cells to H<sub>2</sub>O<sub>2</sub> in serum-free (i.e. transferrin-free) KH buffer, the mechanism proposed by Kotamraju et al. [22], namely inhibition of transferrin receptor-mediated iron uptake by NO, can not contribute to the protection provided by NO. Inhibition of mitochondrial respiration by NO, as suggested to contribute to protection against peroxide-mediated delayed cell death [23], is also unlikely, as in the model employed here and under the oxygen partial pressures (i.e. ambient oxygen) used here, 10  $\mu$ M SpNO did not inhibit the mitochondrial respiratory chain to any appreciable extent [29]. Mediation of protective effects of NO by cGMP and PKG has also been reported in several models of cell injury, e.g. in models of ischemia/reperfusion injury [14,19]. However, the protection of

NO against H<sub>2</sub>O<sub>2</sub> toxicity in our experiments was neither mimicked by 8-Br-cGMP nor inhibited by the PKG inhibitor KT5823. Therefore, this pathway does not appear to be involved in the protection by NO against H<sub>2</sub>O<sub>2</sub> toxicity in our experiments.

The cytotoxicity high NO concentrations elicited in the absence of H<sub>2</sub>O<sub>2</sub> is due to NO<sub>x</sub> as we have shown previously (Figure 10) [29]; the well-known inhibition of cytochrome oxidase by NO, which occurs in competition to oxygen, does not contribute significantly to the endothelial cell injury in our model as experiments were performed at 21% O<sub>2</sub> and as the experiments with the highly glycolytic liver endothelial cells [41] were performed in the presence of glucose [29]. We cannot exclude that NO, which exerts multiple effects in biological systems [2,3,42], might contribute to cytoprotection or to enhancement of H<sub>2</sub>O<sub>2</sub>-induced cell injury by other ways; however, inhibition of lipid peroxidation and inhibition of H<sub>2</sub>O<sub>2</sub> degradation can fully account for the protection against and the enhancement of H<sub>2</sub>O<sub>2</sub> cytotoxicity, respectively, in the model studied here, and can sufficiently explain the at first view contradictory results on the effects of NO on H<sub>2</sub>O<sub>2</sub> toxicity.

Taking into consideration the NO concentrations which are assumed to occur *in vivo* under physiological (well below 1  $\mu$ M NO) and pathophysiological conditions (presumably below 10  $\mu$ M NO) [3,43,44], one should expect that under most conditions, the protective effects of NO prevail. On the other hand, physiological *in vivo* concentrations of H<sub>2</sub>O<sub>2</sub> have been reported to be in the range of 1–100 nM [45], so that under pathological conditions in which elevated amounts of both NO and reactive oxygen species are formed, e.g. during inflammatory reactions or during reperfusion of tissues, inhibition of catalase and/or glutathione peroxidase might also contribute significantly to the injurious process. The balance of these processes is likely to vary considerably depending on the conditions, cell types involved, oxygen partial pressure, etc. so that overall a very delicate balance between enhancing and protective effects of NO is to be expected under these conditions. Furthermore, on the tissue level the vasodilatory actions of NO are, besides direct cellular effects as discussed in this article, likely to contribute to tissue protection [2,46]. In addition, H<sub>2</sub>O<sub>2</sub> has been described to enhance eNOS activity [47], the NO thus released might contribute to the protection against (or the enhancement of) the oxidative stress generated by the H<sub>2</sub>O<sub>2</sub> that stimulates its release.

In conclusion, we have shown here that the inhibition of H<sub>2</sub>O<sub>2</sub> degradation requires relatively high NO concentrations while inhibition of lipid peroxidation only requires low NO concentrations. Thus, at low NO concentrations protection against H<sub>2</sub>O<sub>2</sub> toxicity can be observed. At high NO concentrations, both inhibition of H<sub>2</sub>O<sub>2</sub> degradation

and inhibition of lipid peroxidation can be observed and it appears to be dependent on the relative importance of both processes whether NO enhances or inhibits H<sub>2</sub>O<sub>2</sub>-induced cell injury.

### Acknowledgement

The authors thank Ms M. Holzhauser, Mrs M. Brachvogel and Ms N. Boschenkov for their excellent technical assistance.

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